

New Macrocyclic Derivatives Containing Chiral and Linear Lateral Amino-Acid Moieties

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We report the preparation of a new series of functionalized derivatives of dibenzo-18-crown-6, containing in second sphere of coordination lateral arms of different linear and chiral L-amino acids: glycine, β -alanine, 11-aminoundecanoic acid, L-cysteine, L-aspartic acid, L-phenylalanine, L-

tryptophan, L-histidine. All these new derivatives were characterized by elemental analysis, IR and ¹H-NMR spectroscopy, and these data were in agreement with proposed molecular structures.

Molecular receptors are defined as organic structures held by covalent bonds, that are able to bind selectively ionic or molecular substrates by means of various intermolecular interactions, leading to an assembly of two or more species, a *supermolecule*.^[1] Complexation of ionic substrates^{[1][2][3][4][5][6][7][8]} was achieved by the simplest recognition with synthetic receptors such as crown ethers and their numerous derivatives, cryptates or polyaza- and polythia-macrocyclic ligands. Tetrahedral, trigonal or multiple recognition was extended for selective binding of molecular substrates.^{[9][10]} Because of the important role played by substituted ammonium ions (i.e. amino acids, ephedrine, adrenaline etc.) in chemistry and in biology, the preparation of the optically active crown ethers **1**, **3**^{[11][12]} or the tritopic selective receptor **2**^[10], capable of selectively recognizing of such substrates, is of special interest (Figure 1).

We previously reported the first ditopic receptors containing positively charged moieties in the second sphere of coordination^{[13][14]} for which we expected a new kind of complexation behavior. For instance, amino acids are known to bind to crown ethers^{[4][6][15]}; the ammonium moiety of complexed amino acids is able to interact with a crown ether cavity by three hydrogen bonds. The ditopic receptors are able to interact with the carboxylate anion (from the zwitterionic form of the amino acid) under specific conditions.^{[15][16]} In this paper we describe the prepa-

ration of a new series of functionalized dibenzo-18-crown-6 ether derivatives containing linear or optically active L-amino acid moieties in the second sphere of coordination. The objective is to use these derivatives as polytopic receptors, as mobile carriers in liquid membrane separation, as analytical reagents for spectrophotometric determination of optically active compounds, as well as molecular models for enzymatic catalysis.^{[17][18][19]} The selected macrocyclic receptor used in our syntheses was the commercially available dibenzo-18-crown-6 (**4**). The acylation reaction, using acetic acid, carried out in Eaton reagent (CH₃SO₃H/P₂O₅, 1:12 w/w)^[20] yielded the 2,3,11,12-bis[4(5')-acetyl]benzo-18-crown-6 derivatives **5a,b**.^{[14][20]} Subsequent oxidation of crude **5a,b** with NaBrO/NaOH^[22] led to 2,3,11,12-bis(4-carboxy)benzo-18-crown-6 (**6**; Figure 2).

In the first step, a *cis/trans* mixture of diacylated compounds **5a,b** was obtained, which could not be purified by column chromatography or by fractional crystallization. In general, for similar systems the *cis* isomer is the predominant one.^{[20][21]} Previously, the acylation reaction of benzo-crown ethers with phthalimido-protected amino acids led to the same result.^[15] Oxidation of the *cis/trans* derivatives **5a,b**^[22] afforded a mixture of crude dicarboxylic acids, from which the *cis* product **6** was isolated by fractional crystallization from EtOH. Unfortunately, the *trans*-dicarboxylic acid (15% from crude mixture), could not be satisfactorily purified due to its low solubility. By reaction with SOCl₂, the *cis*-dicarboxylic acid led to the diacyl dichloride deriva-

[◇] Part II: Ref.^[16].

Figure 1. Selective macrocyclic receptors for biologically active ammonium substrates

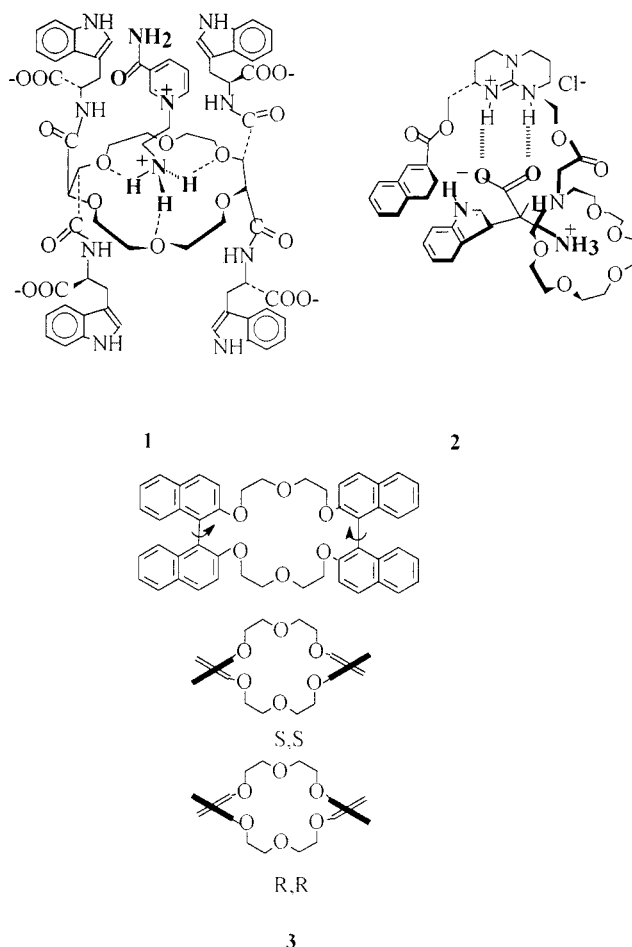
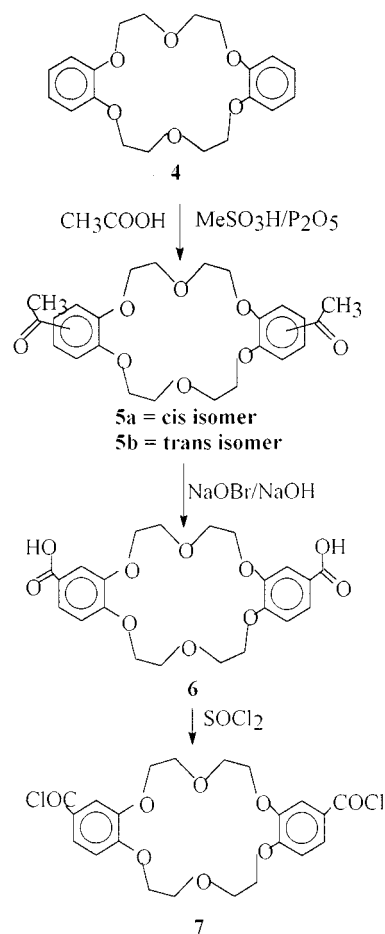


Figure 2. Acylation and oxidation reactions of dibenzo-18-crown-6



tive **7** (see Figure 2) which reacts with glycine (**8**), β -alanine (**9**), 11-aminoundecanoic acid (**10**), L-cysteine (**11**), L-aspartic acid (**12**), L-phenylalanine (**13**), L-tryptophan (**14**), L-histidine (**15**) (see Figure 3), to yield the ditopic macrocyclic derivatives **8–15**.

The new compounds **5–15** were characterized by elemental analysis, IR and ^1H -NMR spectroscopy. The IR spectra of **5a,b** showed a broad ν_{CO} band at $1630\text{--}1674\text{ cm}^{-1}$, probably due to the two isomers. After purification of the crude dicarboxylated compounds, the *cis* compound **6** presented two ν_{CO} bands at 1677 and 1730 cm^{-1} .^[22] The *trans* product also shows two ν_{CO} bands at 1635 and 1730 cm^{-1} . The IR spectra of amino acid derivatives **8–15** showed the characteristic vibrations of the crown ether moiety: $\nu_{\text{COCas}} = 1267\text{--}1270\text{ cm}^{-1}$, $\nu_{\text{COCs}} = 1131\text{--}1134\text{ cm}^{-1}$, $\nu_{\text{COCAr}} = 1062\text{ cm}^{-1}$, $\nu_{\text{CCAr}} = 1500\text{--}1509\text{ cm}^{-1}$, $\nu_{\text{Ar conj.}} = 1588\text{--}1600\text{ cm}^{-1}$, $\nu_{\text{CH}_2\text{s}} = 2875\text{--}2886\text{ cm}^{-1}$, $\nu_{\text{CH}_2\text{as}} = 2923\text{--}2933\text{ cm}^{-1}$ and $\nu_{\text{CHAr}} = 3075\text{--}3086\text{ cm}^{-1}$. The bands of the amide moiety were also detected: $\nu_{\text{NHCO I}} = 1630\text{--}1680\text{ cm}^{-1}$, $\nu_{\text{NHCO II}} = 1451\text{--}1462\text{ cm}^{-1}$ and $\nu_{\text{NHCO III}} = 1341\text{--}1362\text{ cm}^{-1}$. For the linear amino acid derivatives **8–10**, the $\nu_{\text{NHCO I}}$ and $\nu_{\text{NHCO III}}$ bands appeared at $1675\text{--}1682\text{ cm}^{-1}$ and $1356\text{--}1362\text{ cm}^{-1}$, respectively, due to the non-associated amide moieties. For the optically active

amino acid derivatives **11–15** the $\nu_{\text{NHCO II}}$ and $\nu_{\text{NHCO III}}$ bands were displaced to lower frequencies ($\nu_{\text{NHCO II (NH)}}: 1550 \Rightarrow 1451\text{ cm}^{-1}$ and $\nu_{\text{NHCO (CN) III}}: 1400 \Rightarrow 1350\text{ cm}^{-1}$), probably due to intramolecular hydrogen bonding with amide, carboxylic or heterocyclic moieties (Figure 4).^[23]

In the IR spectra of compounds **8–15** the carboxylic bands of the amino acids are present at $\nu_{\text{CO}} = 1710\text{--}1720\text{ cm}^{-1}$ and $\nu_{\text{OH as}} = 3200\text{--}3500\text{ cm}^{-1}$ and $\nu_{\text{OH nonas}} = 3610\text{--}3618\text{ cm}^{-1}$. In the ^1H -NMR spectra of the synthesized compounds the following features were observed:

i) the proton signals of the ether bridges of the crown ether moieties appear as triplets at $\delta = 3.70\text{--}4.30$ for the unsubstituted derivatives **5,6** and at $\delta = 3.83\text{--}4.11$ for the amino acid compounds **8–15**;

ii) the signals of the aromatic protons of the acylated phenyl ring are observed at $\delta = 6.96\text{--}7.00$ and at $\delta = 7.25\text{--}7.51$;

iii) the carboxylic proton signals appear as a singlet at $\delta = 9.94\text{--}10.04$;

iv) in the ^1H -NMR spectra of the chiral and linear amino acid derivatives the proton signals of the R groups were assigned (for details see Experimental Section);

v) in the ^1H -NMR spectra of the amides **8–10** the NH proton signals appear at $\delta = 8.11\text{--}8.28$, and a considerable

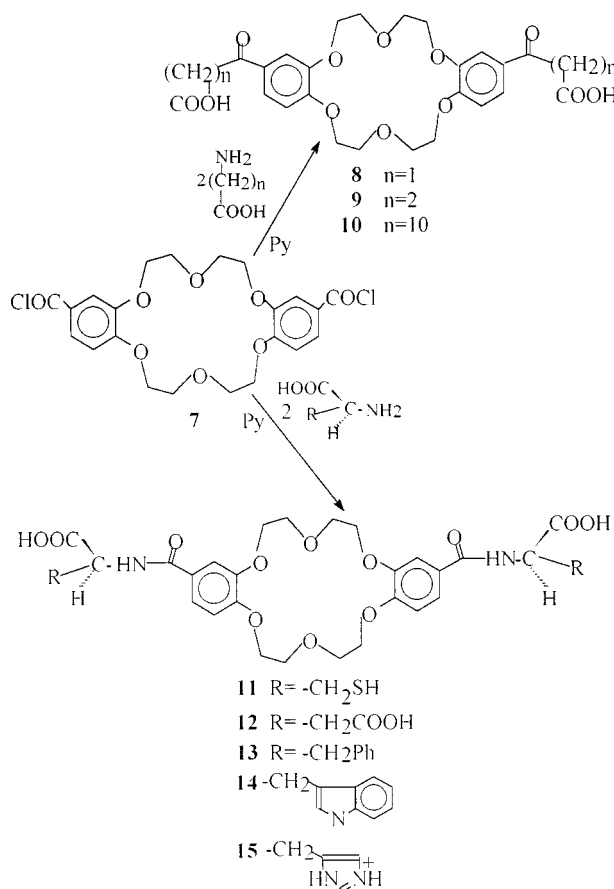
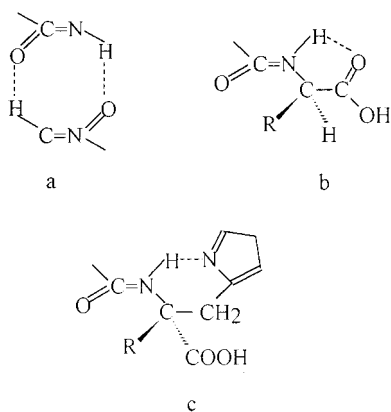
Figure 3. Synthesis of macrocyclic amino acid derivatives **8–15** from diacyl dichloride **7**

Figure 4. Hydrogen bonding of amide moieties with: a) the other amide moiety; b) carboxylic acid, or c) heterocyclic moiety from the amino acid group



downshift is observed for those protons ($\delta = 10.44$ – 10.65) in the derivatives **11–15**,^[24] indicating the formation of hydrogen bonds between amide groups and available moieties of the amino acid groups (see Figure 4).

Further enantioselective transport experiments with the amino acids using liquid membranes are currently under way.

Experimental Section

L-Amino acids, dibenzo-18-crown-6, methanesulfonic acid, and phosphorus pentoxide were purchased from Merck and were used without purification. All other reagents and solvents were of analytical grade and were used as received. – Melting points were determined with a heating-plate microscope and were not corrected. – IR spectra were recorded using KBr pellets with a Nicolet 2DXFT-IR spectrometer. – ^1H -NMR spectra were recorded with a Bruker AC 250-MHz spectrometer in $[\text{D}_6]\text{DMSO}$. Chemical shifts are reported as δ values relative to TMS as internal standard. – Elemental analysis was done by combustion with a Carlo Erba instrument CHNS Elemental Analyser Model 1106.

Preparation of 2,3,11,12-Bis[4(5')-acetyl]benzo-18-crown-6 (5a,b): 3 g ($8.33 \cdot 10^{-3}$ mol) of dibenzo-18-crown-6 (**4**) and 0.7 g (0.7 ml) of acetic acid were added to 15 ml of Eaton reagent.^[20] The reaction mixture was stirred for 48 h at room temp., then for 20 min at 60°C . The reaction mixture was poured into 200 ml of water, causing precipitation of the desired product. After filtration, this was redissolved in 50 ml of CHCl_3 and the obtained solution was washed with diluted LiOH, then with water. The solvent was evaporated in vacuo and the separated products were recrystallized from *n*-hexane (3.33 g, 90%). – M.p. 165 – 167°C from *n*-hexane. – IR (KBr): $\nu = 798, 1058, 1140, 1274, 1360, 1430, 1516, 1598, 1630$ – $1674, 2880, 2932\text{ cm}^{-1}$. – ^1H NMR ($[\text{D}_6]\text{DMSO}$): $\delta = 2.53$ (s, 6 H, CH_3), 3.70 – 4.23 (m, 16 H, CH_2CH_2), 6.82 – 6.86 [m, 2 H, *m*- $\text{H}-\text{C}_6\text{H}_3\text{O}(\text{CO})$], 7.49 – 7.52 [m, 4 H, *o*- $\text{H}-\text{C}_6\text{H}_3\text{O}(\text{CO})$]. – $\text{C}_{24}\text{H}_{28}\text{O}_8$ (438.43): calcd. C 65.75, H 5.06, O 29.19; found C 65.52, H 5.49, O, 28.86.

Preparation of 2,3,11,12-Bis(4-carboxy)benzo-18-crown-6 (6): A mixture of the *cis*- and *trans* isomers of 2,3,11,12-bis[4(5')-acetyl]benzo-18-crown-6 (**5a,b**) (3 g, $6.75 \cdot 10^{-3}$ mol), was suspended in 100 ml of distilled water, and 150 ml of an aqueous solutions of NaOH (44 g) and 14 ml of bromine water (276 mmol Br_2) were added at 0 – 10°C within 2 h; then the solution was stirred at 20°C for 24 h. After extraction of unreacted **5a,b** with chloroform (200 ml), the aqueous solution was acidified with hydrochloric acid (2 N). The product precipitated. Recrystallization from EtOH led to the *cis* isomer of 2,3,11,12-bis[4(5')-carboxy]benzo-18-crown-6 (**6**) (2.42 g, 80%), whereas the insoluble residue contained a small quantity of the *trans* product.

General Procedure for the Preparation of the Amino Acid Derivatives 8–15: 0.9 g ($2 \cdot 10^{-3}$ mol) of dicarboxylic acid **6** was suspended in 100 ml of freshly distilled benzene. Then 30 mmol of SOCl_2 was added and the reaction mixture was refluxed for 5 h. After cooling, the solvents were removed under vacuum and the crude diacyl dichloride **7** was dissolved in 50 ml of dry pyridine. A stoichiometric amount of the amino acid was added at 0 – 10°C during 30 min, and then the reaction mixture was refluxed for 1 h. The reaction mixture was poured into 200 ml of ice-cold water and the resulting product [0.92 g of **8** (80%), 0.94 g of **9** (79%), 1.14 g of **10** (70%), 0.94 g of **11** (72%), 0.82 g of **12** (60%), 1.19 g of **13** (75%), 1.13 g of **14** (78%), 1.32 g of **15** (80%),] was recrystallized from acetic acid.

2,3,11,12-Bis[4-carboxyl]benzo-18-crown-6 (6): M.p. $> 250^\circ\text{C}$ from EtOH. – IR (KBr): $\nu = 884, 957, 1067, 1131, 1204, 1267, 1367, 1456, 1504, 1598, 1677, 1730, 2875, 2928, 3432, 3621\text{ cm}^{-1}$. – ^1H NMR ($[\text{D}_6]\text{DMSO}$): $\delta = 3.82$ – 4.20 (m, 16 H, CH_2CH_2), 7.02 – 7.06 [m, 2 H, *m*- $\text{H}-\text{C}_6\text{H}_3\text{O}(\text{CO})$], 7.20 – 7.54 [m, 4 H, *o*- $\text{H}-\text{C}_6\text{H}_3\text{O}(\text{CO})$], 10.04 (s, 2 H, COOH). – $\text{C}_{22}\text{H}_{24}\text{O}_{10}$ (448.43): calcd. C 58.93, H 5.39, O 35.68; found C 58.65, H 5.49, O 35.77.

Glycine Derivative 8: M.p. 195 – 197°C from AcOH. – IR (KBr): $\nu = 875, 960, 1060, 1132, 1204, 1270, 1361, 1452, 1500, 1600, 1680, 1720, 1750, 2883, 2885, 2924, 3080, 3400, 3610\text{ cm}^{-1}$. – ^1H NMR ($[\text{D}_6]\text{DMSO}$): $\delta = 3.88$ – 4.11 (m, 16 H, CH_2CH_2), 5.60 (m, 4 H,

HNCH₂COOH), 7.00 [m, 2 H, *m*-H-C₆H₃O(CO)], 7.21–7.52 [m, 4 H, *o*-H-C₆H₃O(CO)], 8.28 (m, 2 H, HNCO), 10.01 (s, 2 H, COOH). – C₂₆H₃₀N₂O₁₂ (562.53): calcd. C 55.51, H 5.38, N 4.98, O 34.13; found C 55.16; H 5.79, N 4.91, O 34.16.

β-Alanine Derivative **9**: M.p. 198–198°C from AcOH. – IR (KBr): ν = 878, 957, 1062, 1131, 1204, 1272, 1362, 1436, 1451, 1504, 1588, 1688, 1698, 2886, 2923, 3086, 3422, 3620 cm⁻¹. – ¹H NMR ([D₆]DMSO): δ = 2.15 (t, 4 H, CH₂COOH), 3.18 (t, 4 H, CH₂NHCO), 3.88–4.11 (m, 16 H, CH₂CH₂), 7.01 [m, 2 H, *m*-H-C₆H₃O(CO)], 7.20–7.56 [m, 4 H, *o*-H-C₆H₃O(CO)], 8.29 (m, 2 H, HNCO), 10.06 (s, 2 H, COOH). – C₂₆H₃₀N₂O₁₂ (562.53): calcd. C 55.51, H 5.38, N 4.98, O 34.13; found C 55.16, H 5.79, N 4.91, O 34.16.

11-Aminoundecanoic Acid Derivative 10: M.p. 135–137°C from AcOH. – IR (KBr): ν = 757, 878, 1062, 1136, 1204, 1267, 1356, 1451, 1504, 1588, 1640, 1682, 1724, 2602, 2849, 2933, 3080, 3306, 3411, 3616 cm⁻¹. – ¹H NMR ([D₆]DMSO): δ = 1.24–1.47 [m, 32 H, (CH₂)₈], 2.14 (t, 4 H, CH₂COOH), 3.23 (t, 4 H, CH₂NHCO), 3.82–4.34 (m, 16 H, CH₂CH₂), 6.96 [m, 2 H, *m*-H-C₆H₃O(CO)], 7.25–7.51 [m, 4 H, *o*-H-C₆H₃O(CO)], 8.28 (m, 2 H, HNCO), 10.05 (s, 2 H, COOH). – C₄₄H₆₆N₂O₁₂ (815.01): calcd. C 64.84, H 8.16, N 3.44, O 23.56; found C 64.58, H 8.39, N 3.51, O 23.16.

L-Cysteine Derivative 11: M.p. 188–189°C from AcOH. – IR (KBr): ν = 768, 878, 1062, 1136, 1204, 1267, 1351, 1430, 1456, 1504, 1604, 1635, 1720, 1750, 2883, 2885, 2924, 3080, 3400, 3610 cm⁻¹. – ¹H NMR ([D₆]DMSO): δ = 2.61 (d, 4 H, CH₂SH), 3.78 [t, 2 H, HOOCCH(R)NHCO], 3.82–4.11 (m, 16 H, CH₂CH₂), 7.04 [m, 2 H, *m*-H-C₆H₃O(CO)], 7.24–7.75 [m, 4 H, *o*-H-C₆H₃O(CO)], 10.02 (s, 2 H, COOH), 10.44 (m, 2 H, HNCO). – C₂₆H₃₀N₂O₁₂S (650.67): calcd. C 51.69, H 4.65, N 4.31, O 29.51, S 9.85; found C 51.32, H 4.79, N 4.14, O 29.44, S 9.99.

L-Aspartic Acid Derivative 12: M.p. 220–222°C from AcOH. – IR (KBr): ν = 873, 962, 1062, 1131, 1204, 1272, 1350, 1451, 1514, 1598, 1682, 1714, 1756, 2613, 2870, 2923, 3080, 3442, 3627 cm⁻¹. – ¹H NMR ([D₆]DMSO): δ = 2.39 (d, 4 H, CH₂COOH), 3.8 [m, 2 H, HOOCCH(R)NHCO], 3.89–4.11 (m, 16 H, CH₂CH₂), 7.00 [m, 2 H, *m*-H-C₆H₃O(CO)], 7.20–7.66 [m, 4 H, *o*-H-C₆H₃O(CO)], 10.26 (s, 2 H, COOH), 10.47 (s, 2 H, COOH), 10.58 (m, 2 H, HNCO). – C₃₀H₃₄N₂O₁₆ (678.60): calcd. C 53.10, H 5.05, N 4.13, O 37.72; found C 53.35, H 4.95, N 4.00, O 37.64.

L-Phenylalanine Derivative 13: M.p. 185–187°C from AcOH. – IR (KBr): ν = 705, 747, 957, 1062, 1131, 1204, 1267, 1341, 1451, 1498, 1604, 1640, 1709, 2108, 2592, 2881, 2928, 3028, 3075, 3411, 3610 cm⁻¹. – ¹H NMR ([D₆]DMSO): δ = 3.15 (d, 4 H, CH₂C₆H₅), 3.78 [m, 2 H, HOOCCH(R)NHCO], 3.83–4.11 (m, 16 H, CH₂CH₂), 7.01 [m, 2 H, *m*-H-C₆H₃O(CO)], 7.21–7.57 [m, 16 H, *o*-H-C₆H₃O(CO) and C₆H₅CH₂], 10.02 (s, 2 H, COOH), 10.44 (m, 2 H, HNCO). – C₄₀H₄₂N₂O₁₂ (742.78): calcd. C 64.64, H 5.70, N 4.13, O 25.85; found C 64.22, H 5.91, N 4.02, O 25.65.

L-Histidine Derivative 14: M.p. 177–179°C from AcOH. – IR (KBr): ν = 884, 957, 1062, 1136, 1204, 1272, 1360, 1435, 1462, 1509, 1593, 1680, 1709, 1740, 2602, 2875, 2933, 3075, 3417, 3632 cm⁻¹. – ¹H NMR ([D₆]DMSO): δ = 3.06 (d, 4 H, CH₂R), 3.78 [m, 2 H, HOOCCH(R)NHCO], 3.88–4.13 (m, 16 H, CH₂CH₂),

7.02 [m, 2 H, *m*-H-C₆H₃O(CO)], 7.21–7.70 [m, 12 H, *o*-H-C₆H₃O(CO) and the aromatic protons of histidine residues], 8.44 (m, 2 H, HNCO), 9.94 (s, 2 H, COOH). – C₃₄H₃₈N₆O₁₂ (722.71): calcd. C 56.51, H 5.30, N 11.63, O 26.57; found C 56.89, H 5.11, N 11.35, O 25.38.

L-Tryptophan Derivative 15: M.p. 181–183°C from AcOH. – IR (KBr): ν = 873, 962, 1062, 1136, 1204, 1267, 1341, 1435, 1456, 1504, 1598, 1640, 1703, 2613, 2875, 2928, 3080, 3243, 3411, 3616 cm⁻¹. – ¹H NMR ([D₆]DMSO): δ = 3.37 (d, 4 H, CH₂Ar), 3.78 [m, 2 H, HOOCCH(R)NHCO], 3.83–4.11 (m, 16 H, CH₂CH₂), 6.44–7.01 [m, 4 H, *m*-H-C₆H₃O(CO) and the aromatic protons of tryptophan moieties], 7.10–7.57 [m, 14 H, *o*-H-C₆H₃O(CO) and the aromatic protons of tryptophan moieties], 10.02 (s, 2 H, COOH), 10.1 (m, 2 H, HN of tryptophan moieties), 10.54 (m, 2 H, HNCO). – C₄₄H₆₆N₄O₁₂ (843.03): calcd. C 62.69; H 7.89, N 6.65, O 22.77; found C 62.32, H 7.91, N 6.82, O 23.00.

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